

¹H NMR SPECTRAL EVIDENCE FOR
THE STRUCTURE AND CONFORMATION
OF PEPTIDE ANTIBIOTIC SIOMYCIN-A

Sir:

The chemical structures of the peptide antibiotic siomycin-A (**I**) isolated from *Streptomyces sioyaensis*¹⁾ as well as thiostrepton (**II**) produced by *S. azureus*²⁾ had earlier been studied by chemical degradations^{3,4)}. In 1970, much of the structure of **II** was determined by an X-ray crystallographic analysis⁵⁾. In 1976, we proposed the total structure of **II** on the basis of the X-ray analysis⁵⁾ and 15-MHz ¹³C NMR spectroscopy⁶⁾, and also the total structure of **I** on the basis of the close relationships, between **I** and **II**, of the degradation products^{3,4,7)} and ¹³C NMR spectra⁶⁾. We further confirmed the structure relationship between **I** and **II** by the analysis of the 36.5-MHz ¹⁵N NMR spectra⁸⁾.

However, the correlations between the structures and ¹H NMR spectra of **I** and **II** have not yet been elucidated in detail, which should provide some useful information about their stereostructures and conformations in solution. Recently the structures of similar antibiotics thiopeptins⁹⁾ were proposed from studies by 300-MHz ¹H and 75-MHz ¹³C NMR spectroscopy¹⁰⁾. We report here the detailed analyses of 270-MHz ¹H NMR spectra of **I**, **II**, and thiopeptin-B₂ (**III**), which was isolated from commercially available thiopeptin mixtures⁹⁾, and propose the molecular conformation of these antibiotics in solution.

The ¹H signals were assigned by detailed spin-decoupling experiments and by comparisons of chemical shifts in various solvents. The OH and NH signals were revealed by the addition of a small amount of D₂O to CDCl₃ solutions. Additions of CD₃OD or CD₃COOD to CDCl₃ solutions did not completely exchange some of CONH protons by deuterons (see Fig. 1). All CONH protons were exchanged in pure CD₃-COOD. However, **I** rapidly decomposed by the addition of CF₃COOD. The signal assignments with chemical shift data are listed in Table 1 and the coupling constants, J values, in Table 2.

Intramolecular interproton OVERHAUSER effects¹¹⁾ were measured in a degassed CDCl₃ solution of **I** (at 23°C and 60°C), by observing the difference spectra between the selective on-resonance gated-¹H-irradiated (with NOE) and off-resonance gated-¹H-irradiated (without NOE) modes¹²⁾. In preliminary experiments, about 25% positive NOE enhancement was observed at 100-MHz, between the two geminal =CH₂ signals of a dehydroalanine (Deala) residue of **I** in CDCl₃ at 30°C, although precise NOE measurements were not possible because of signal overlappings. However, the correlation time was roughly estimated as 6~8 × 10⁻¹⁰ sec by 25-MHz ¹³C NMR spectroscopy¹³⁾. Accordingly, at 270-MHz, the NOE enhancements of proton resonances of **I** in CDCl₃ were much smaller (several % at most) and negative at 23°C, as expected for large globular molecules^{14,15)}. Therefore, we also measured the NOE

Fig. 1. 270-MHz ¹H NMR spectra of siomycin-A (recrystallized from CHCl₃-CH₃OH) in CDCl₃ (bottom) and siomycin-A (recrystallized from CDCl₃-CD₃OD) in CDCl₃-D₂O (8:1) (upper trace).

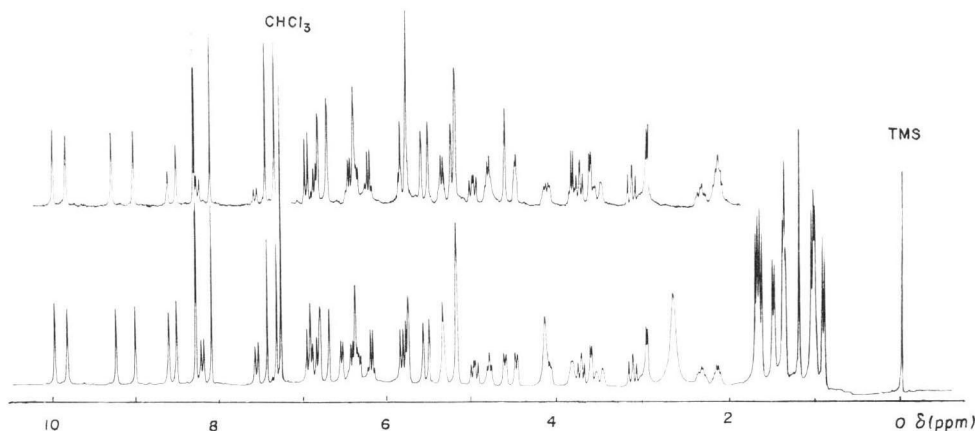


Table 1. ¹H Chemical shift data^a.

Assignments	I			II		III		
	CDCl ₃	{CDCl ₃ CD ₃ OD}	CD ₃ CO ₂ D	CDCl ₃	CD ₃ CO ₂ D	CDCl ₃	CD ₃ CO ₂ D	
Ala-1	β CH ₃ αCH CONH	1.48d 4.80dq 6.54bd	1.47d 4.81dq 7.35bd ^b	1.42d 4.87q —	1.46d 4.77dq 6.47bd	1.44d 4.81q —	1.54d 4.96dq 6.85bd	1.63d 4.98q —
Deala-1	β =CH(c) β =CH(t) CONH	5.18bs 5.76d 8.60bs	5.42bs 5.80d 8.64bs ^b	5.62bs 5.85bs —	5.21bs 5.78d 7.80bs	5.61bs 5.94bs —	5.26bs 5.82d 7.87bs	5.64bs 5.94bs —
Deala-2	β =CH(c) β =CH(t) CONH	5.18bs 6.38bs 9.23bs	5.17bs 6.39bs 9.37bs ^b	5.30bs 6.48bs —	— — —	— — —	— — —	— — —
Ala-2	β CH ₃ αCH CONH	— — —	— — —	— — —	1.20d 3.83dq 7.58bd	1.19d 4.35q —	1.22d 3.87dq 7.65bd	1.20d 4.35d —
Val	γ CH ₃ β CH αCH	0.91d 1.04d 2.12m 2.94d	0.81d 1.10d 2.37m 3.01d	0.83d 1.09d 2.35m 3.22d	— — — —	— — — —	0.95d 0.99d 1.94m 2.90d	0.78d 1.11d 2.39m 3.31d
Ile	γ CH ₃ γ CH ₃ γ CH ₂ β CH αCH	— — — — —	— — — — —	— — — — —	0.95t 0.90d ~1.3m ~1.7m 2.99d	0.89t 1.08d ~1.6m ~2.1m 3.34d	— — — — —	— — — — —
Q	3=CH 5=CH 6=CH 7=CH 8=CH 11=CH 11-CH ₃ 8-OH	7.31s 6.94d 6.35dd 3.61d 4.61d 5.34bq 1.37d 6.91d	7.31s 6.94d 6.44dd 3.60d 4.42s 5.34q 1.40d —	7.41s 6.98d 6.51dd 3.69d 4.69s 5.47q 1.42d —	7.31s 6.89d 6.31dd 3.63d 4.68d 5.33bq 1.37d 6.89d	7.47s 6.97d 6.51dd 3.72d 4.64s 5.47q 1.44d —	7.36s 6.92d 6.34dd 3.62d 4.67d 5.37bq 1.37d 6.85d	7.47s 6.96d 6.51dd 3.69d 4.60s 5.47q 1.43d —
Thr-2	γ CH ₃ β CH αCH CONH	1.68d 6.43q 5.83d 8.19bd	1.69d 6.37q 5.80d 8.73bd ^b	1.68d 6.51q 5.94s —	1.74d 6.39q 5.83d 8.32bd	1.69d 6.51q 5.94bs 7.96bd ^b	1.77d 6.56q 6.81d 9.80bd	1.67d 6.60q 6.77s 9.63bd ^b
Thstn	γ CH ₃ γ CH β CH ₃ αCH CONH Thz-4 =CH	1.35d 3.83bq 1.19s 5.77d 7.55bd 8.27s	1.32d 3.82q 1.18s 5.78d 7.58bd 8.31s	1.31d 3.98q 1.19s 5.84s — 8.55s	1.34d 3.82bq 1.19s 5.76d 7.57bd 8.28s	1.42d 4.00q 1.26s 5.85s 7.74bd ^b 8.56s	1.37d 3.81bq 1.21s 5.77d 7.56bd 8.56s ^e	1.41d 3.99q 1.24s 5.89s 8.10bd ^b 8.62s ^e
(+)-Cys	β CH β' CH αCH	3.12dd 3.72dd 4.96dd	3.18dd 3.66dd 5.00dd	3.25dd 3.71dd 5.17dd	3.11dd 3.70dd 4.95dd	3.25dd 3.72dd 5.19dd	3.11dd 3.71dd 4.94dd	3.24dd 3.71dd 5.15dd
Debut	γ CH ₃ β CH CONH	1.62d 6.19q 8.51bs	1.64d 6.23q 8.55bs	1.65d 6.30q —	1.63d 6.19q 8.50bs	1.64d 6.30q —	1.62d 6.20q 8.48bs	1.65d 6.28q —
Thr-1	γ CH ₃ β CH αCH CONH	1.01d ^e ~1.25m 4.48bd 6.83bd	0.86d 1.57dq 4.43dd 7.09bd	1.04d ~1.60m 4.58d —	1.00d ^e ~1.05m 4.45bd 6.84bd	1.06d ^e ~1.30m 4.64d 7.17bd ^b	0.97d ^e ~1.10m 4.44bd 6.87bd	0.96d ~1.30m 4.58d —
ThstA ^d	P-2=CH P-3a=CH P-3e=CH P-4a=CH P-4e=CH P-6a=CH	— 2.97dddd 3.51dddd 2.30ddd 4.11ddd 5.18bs	— 2.96dddd 3.51dddd 2.33ddd 4.10ddd 5.32bs	— ^h ^h 2.53d 4.15d 5.49s	— 2.96m 3.48m 2.28ddd 4.09ddd 5.14bs	— ^h ^h 2.54d 4.16d 5.49s	4.15dd 2.13m ⁱ 2.13m ⁱ 1.48m ⁱ 2.34m ⁱ 4.46bs	4.35m 2.57m ^j 2.57m ^j — 2.67m ^j 5.28bs

(to be continued)

Table 1. (continued)

Assignments	I			II		III	
	CDCl ₃	$\left\{ \begin{array}{l} \text{CDCl}_3 \\ \text{CD}_3\text{OD} \end{array} \right.$	CD ₃ CO ₂ D	CDCl ₃	CD ₃ CO ₂ D	CDCl ₃	CD ₃ CO ₂ D
CONH	9.81bs	9.81bs	—	9.83bs	—	9.76bs ^g	—
Thz-1 =CH	8.26s	8.30s	8.52s	8.27s	8.53s	8.18s ^e	8.51s ^e
Thz-2 =CH	8.08s	8.17s	8.43s	8.11s	8.42s	8.12s	8.43s
Thz-3 =CH	7.43s	7.57s	7.69s	7.44s	7.68s	7.40s	7.70bs
Deala-S-1	β -CH(c)	5.58bs	5.63d ^f	5.74bs	5.57bs	5.72bs	5.57bs
	β -CH(t)	6.80d	6.71d	6.75bs	6.79bs	6.76bs	6.74bs
	CONH	9.97bs	9.98bs ^b	—	9.96bs	—	9.89bs ^g
Deala-S-2	β -CH(c)	5.51bs	5.71d ^f	5.83bs	5.48bs	5.84bs	6.13bs
	β -CH(t)	6.69d	6.53d	6.50bs	6.69bs	6.52bs	6.74bs
	CONH	8.99bs	9.14bs ^b	—	8.99bs	—	8.69bs

^a The samples were dissolved in CDCl₃, CDCl₃-CD₃OD (4:1), and CD₃COOD (*ca.* 10 mM) [not soluble in water or CD₃OD]. The 270-MHz ¹H NMR spectra of sample solutions at 23°C were measured by a Bruker WH-270 spectrometer equipped with a Nicolet 1180 computer system. The experimental conditions for the pulse FT measurements were: spectral width, 3,600 Hz; number of data points, 16K; flip angle, 90°; acquisition time, 2.26 sec; number of transients, 64~256. Chemical shifts (δ) were measured from the internal standard of tetramethylsilane. Data on II and III in CDCl₃-CD₃OD (4:1) are not described here.

Abbreviations are as follows: Deala, dehydroalanine; Debut, dehydrobutyrine; P, piperidine ring; Q, quinaldic acid precursor⁶⁾; Thstn, thiostreptine residue⁶⁾; ThstA, thiostreptonic acid unit⁶⁾; Thz, thiazole ring; for signal multiplicities: s, singlet; d, doublet; t, triplet; q, quartet; m, multiplet; b, broad signal.

^b Signal intensities are reduced owing to partial deuterium exchanges.

^c Second-order patterns.

^d Abbreviations a and e are axial and equatorial, respectively.

^{e-g} Assignments may be interchanged.

^h Not observed because of the deuterium exchanges through the imine-enamine tautomerism⁹⁾.

^{i-k} Not assigned.

Table 2. ¹H, ¹H Coupling constants (J in Hz) for I in CDCl₃*.

Ala-1	(α CH, CONH)=9.3
Val	(α CH, β CH)=4.0
Q	(5-CH, 6-CH)=9.7; (6-CH, 7-CH)=5.8; (7-CH, 8-CH)=1.8; (8-CH, 8-OH)=7.0
Thr-2	(α CH, β CH)<0.5; (α CH, CONH)=9.3
Thstn	(α CH, CONH)=10.2
(+)-Cys	(α CH, β CH)=11.5; (α CH, β' CH)=9.3; (β CH, β' CH)=-12.8
Thr-1	(α CH, β CH)=1.5; (α CH, CONH)=10.7
ThstA-P	(3a-CH, 3e-CH)=-19.0; (4a-CH, 4e-CH)=-12.5; (3a-CH, 4a-CH)=12.5; (3a-CH, 4e-CH)=6.0; (3e-CH, 4a-CH)=6.0; (3e-CH, 4e-CH)=1.5; (3a-CH, 6a-CH) \approx 1.5; (3e-CH, 6a-CH) \approx 0.5
Deala-S-1	[β -CH(c), β -CH(t)]=-2.2
Deala-S-2	[β -CH(c), β -CH(t)]=-2.5

* About ± 0.5 Hz. All ³J(CH₃, CH) values were about 6.6~7.2 Hz.

difference spectra of I in CDCl₃ at 60°C and, in fact, obtained all NOE's as positive enhancements at 270-MHz (25 and 15% NOE between the geminal =CH₂ of the Deala-S-1 residue), as shown in Table 3. These NOE observations were useful for establishing the signal assign-

ments (Table 1) and further for elucidating the stereostructures as described below.

For investigating the correlations between the stereostructures and ¹H spectra, we constructed molecular models (CPK and Fieser models) on the basis of the stereographic drawing of II (Fig.

Table 3. Positive NOE enhancements observed for I in CDCl₃ at 60°C*.

¹ H signal		NOE (%)	¹ H signal		NOE (%)
Irradiated	Observed		Irradiated	Observed	
Ala-1	αCH	Thr-2 αCH	Debut CONH	Thr-1 αCH	slight
	αCH	ThstA CONH	Thr-1 αCH	Debut CONH	5
Deala-1	β=CH(c)	Q 3=CH	βCH	Thr-2 CONH	10
	β=CH(c)	ThstA Thz-2 =CH	γCH ₃	Debut CONH	slight
Deala-2	CONH	Deala-2 β=CH(c)	γCH ₃	Thr-1 CONH	slight
	β=CH(c)	Deala-1 CONH	CONH	Thr-1 βCH	5
Q	5=CH	Q 11-CH	CONH	ThstA Thz-3 =CH	5
	7-CH	Val αCH	CONH	ThstA P-6a-CH	15
	8-CH	αCH	P-6a-CH	Thz-3 =CH	10
	11-CH	Q 3=CH	CONH	Ala-1 αCH	5
	11-CH	5=CH	Thz-3 =CH	ThstA P-6a-CH	slight
	11-CH	ThstA Thz-2 =CH	Deala-S-2 β=CH(c)	Deala-S-1 CONH	10
	11-CH ₃	Q 3=CH	CONH	β=CH(c)	5

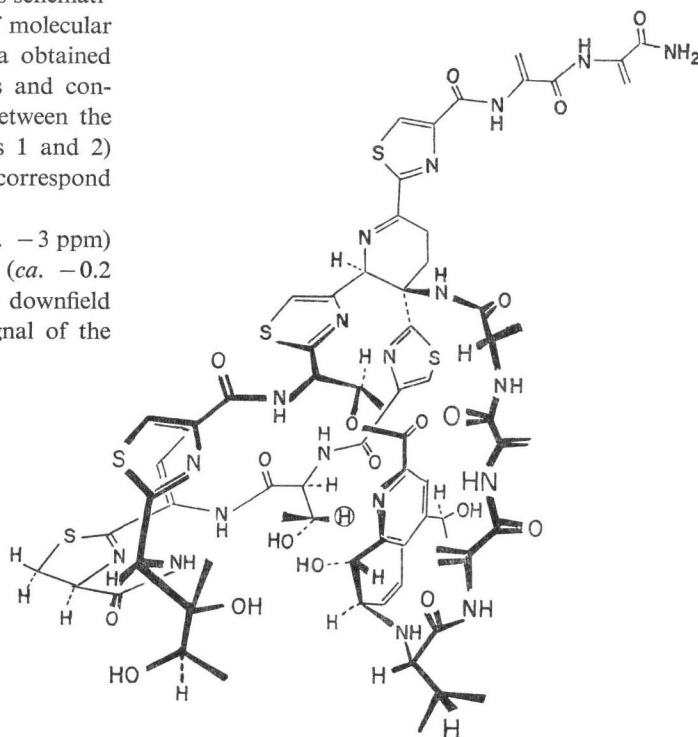
* Positive NOE's were observed between all mutually geminal and all mutually vicinal proton signals except those between mutually vicinal αCH and CONH signals, and are not described here.

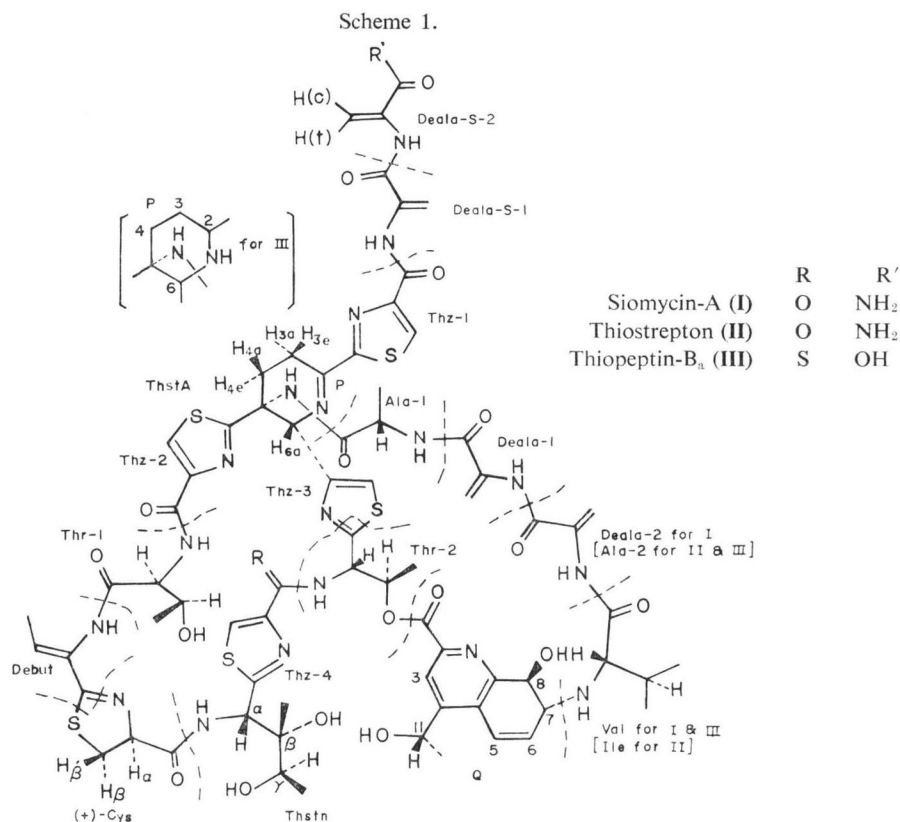
The experimental conditions for NOE measurements were: ¹H-irradiation time, 1 sec; acquisition time, 0.56 sec; total number of transients, 256.

4 of Ref. 5), the dependence of ³J_{H,H} (Table 2) upon dihedral angles¹⁶⁾, and NOE enhancements (Table 3). The molecular model of I is schematically shown in Fig. 2. Inspections of molecular models show that the ¹H NMR data obtained are all consistent with the structures and conformations. Detailed comparisons between the NMR data on I, II, and III (Tables 1 and 2) indicate that their spectral differences correspond to their structural ones.

The extraordinary upfield shift (*ca.* -3 ppm) of the βCH signal, the upfield shift (*ca.* -0.2 ppm) of the γCH₃ signal and the downfield shift (*ca.* +0.4 ppm) of the αCH signal of the Thr-1 residue from the ordinary positions¹⁷⁾ are satisfactorily explained by the ring-current anisotropy effects of the pyridine ring of the quinaldic-acid precursor⁶⁾ (Q) unit; the βCH proton is situated directly above the plane of the pyridine ring (see the circled H in Fig. 2). The pyridine ring also exerts the ring-current effects upon the β=CH signals of the Deala-1 and Deala-2 residues; these β=CH signals appear at appreciably higher fields than do those of the Deala-S-1 and Deala-

Fig. 2. A schematic view of the molecular conformation of siomycin-A.





S-2 residues in the side chain. Some downfield shifts of the Thr-2 signals are also due to the ring-current effects of the pyridine and the thiazole-3 (Thz-3) ring.

In conclusion, the 270-MHz ¹H NMR spectra of I and II were found to be consistent with the total structures proposed previously^{6,8)}. Furthermore, the conformations of the cyclic parts of these antibiotics in solution were found to be similar to the conformation of II in crystal⁵⁾.

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